## Molecular Dynamics and Accelerated Molecular Dynamics

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Lecture 3

# **Overall Outline**

- Day 1: Molecular dynamics methods
- Day 2: Molecular dynamics methods
- Day 3: (accuracy of MD), Infrequent events, transition state theory, saddle finding, etc.
- Day 4: Accelerated molecular dynamics methods

## **Trajectory Accuracy**

Hamiltonian systems, with very few exceptions, are chaotic.

Two trajectories differing initially by an infinitessimal amount  $(\Delta x)$  will diverge exponentially in time -- Lyapunov instability:

 $\Delta \mathbf{x}(t) = \Delta \mathbf{x}(0) \exp(\lambda t)$ 

Thus, any imperfect integrator (and all are imperfect) introduces errors that guarantee the trajectory diverges from the true trajectory.

---> How can we know that we are generating the correct results in MD?

#### Example of chaotic divergence

"cos" potential: 
$$V_{cos}(x,y) = \cos(2\pi x)(1+d_1y) + d_2\frac{(2\pi)^2}{2}y^2$$

## **Trajectory Accuracy**

How can we know that we are generating the correct results in MD?

Answer: We can't know for sure, but there is hope...

#### **Shadow Orbits**

In some cases, there is actually a "shadow orbit" that closely follows the integrated trajectory.



The shadow orbit is an *exact* trajectory for the system, but one that starts from a slightly displaced initial point.

These are known to exist for hyperbolic systems.

Can sometimes be shown to exist, for long times, for more general systems [e.g., see Quinlan and Tremaine, Mon. Not. R. Astron. Soc. **259**, 5050 (1992)].

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#### Deposition event takes ~2 ps – use molecular dynamics (can reach ns)

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- diffusion events affect the film morphology
- mechanisms can be surprisingly complex
- --> need another approach to treat these



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# Direct MD example: Ag/Ag(100)

T=300K MD for 1 ps shown Time to next hop event = 9  $\mu$ s (weeks of CPU time for this very small system)

Ag/Ag(100), EAM potential 55 moving atoms Langevin thermostat

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#### Infrequent Event System



The system vibrates in 3N-dimensional basin.

Occasionally it escapes, crossing through a dividing surface to a new basin. This behavior characterizes solid-state diffusion, as well as many other processes.

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#### Computing the rate for a known mechanism

If initial and final states are known, as well as the reaction coordinate, we can use transition state theory (TST) to compute the rate.



initial state

saddle point (rxn coord shown)

final state

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# Many-dimensional reaction coordinate

The potential energy basin is 3N-dimensional. The reaction coordinate (and hence the dividing surface) can involve multiple atoms.

Example - exchange event on Ag(100) (barrier = 0.64 eV, rxn time =  $\sim$ 1 ms at T=300K)

(Feibelman, 1990)

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#### Transition State Theory (TST)



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TST escape rate = equilibrium flux through dividing surface at x=q

$$k_{A \to B}^{TST} = \langle \delta(x - q) | \dot{x} | \rangle \quad \text{(exact flux)}$$
$$k_{A \to B}^{HTST} = v_0 e^{-\Delta E/k_B T} \quad \text{(harmonic approx.)}$$

- classically exact rate if no recrossings or correlated events
- no dynamics required
- very good approximation for materials diffusion

#### Canonical ensemble average

System in canonical ensemble has a fixed number of atoms (N), volume (V), and temperature (T)

Definition of ensemble average for some property P:

$$\langle P \rangle = \frac{\int_{-\infty}^{\infty} \int_{-\infty}^{\infty} P e^{-\beta H} dx \, dp}{\int_{-\infty}^{\infty} \int_{-\infty}^{\infty} e^{-\beta H} dx \, dp}$$

$$H = V(x) + p^2/2m$$

$$\beta = \frac{1}{k_B T}$$

#### Transition state theory - canonical ensemble

Because Hamiltonian is separable (H = p stuff + x stuff), we can integrate out the momentum part

$$k_{A \to}^{\rm TST} = \left\langle \left| dx/dt \right| \delta(x-q) \right\rangle_A$$



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#### Transition State Theory - generality

TST is very general -- simply based on flux through D.S. No requirement that the bottleneck be an energy barrier. Entropic bottleneck:









## Monte Carlo transition state theory

We can evaluate  $<\delta(x-q)>$  exactly using Metropolis Monte Carlo



- TST box width = w

 $<\delta(x-q)> = (1/2) (1/w) (N_{in}/N_{tot})$ 

 $N_{in}$  = steps inside TST "box"  $N_{tot}$  = steps anywhere in state A

Metropolis walk of adatom and top few layers of substrate

Steps attempting to leave state A are rejected

Extrapolate to w=0

 $k^{TST} = [2k_BT/\pi m]^{1/2} < \delta(x-q) >$  is exact TST escape rate at temperature T (with statistical error bars)

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## **Importance Sampling**

A simple Metropolis walk at low T will rarely reach the TST box



## **Importance Sampling**

One way to improve statistics:



Include displacement vector, +d and -d, in attempted Metropolis steps

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One way to improve statistics:



Include displacement vector, +d and -d, in attempted Metropolis steps [AFV, J. Chem. Phys. 82, 1890 (1985)]

Still a valid Metropolis walk, but much more efficient

 $<\delta(x-q)> = (1/2) (1/w) (N_{in}/N_{tot}) \exp(-\Delta E_{shift}/kT)$ 

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(Keck, Anderson, Bennett, ... 1960's-1970's)

• Use saddle-point trajectories to correct TST



Follow for time  $\tau_{corr}$ , see where they land

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D. Chandler, J. Chem. Phys 68, 2959 (1978):

$$\begin{aligned} k_{A \to B}^{\text{exact}} &= \frac{\langle v_A(0) \delta_A(0) \theta_B(t) \rangle}{\langle |v_A(0)| \, \delta_A(0) \theta_B(0+) \rangle} k_{A \to}^{\text{TST}} & \text{Key requirement:} \\ &= k_{A \to}^{\text{TST}} \frac{2}{N} \sum_{I}^{N} \gamma(I) \theta_B(I, t) & \text{``separation of time scales''} \\ &= k_{A \to}^{\text{TST}} f_d(t) & \tau_{\text{rxn}} >> t > \tau_{\text{corr}} \end{aligned}$$

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# Dynamical corrections theory (cont.)

Neat property:

$$k_{exact} = k^{TST} f_d$$

regardless of position of TST dividing surface!





"Variational TST" (Garrett and Truhlar) - vary dividing surface position to minimize k<sup>TST</sup> (exploiting fact that k<sup>TST</sup> >= k<sup>exact</sup>)
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## Harmonic TST

One-dimensional system; assume potential is perfectly harmonic



#### Harmonic TST - derivation

$$k_{A\to}^{\rm TST} = \left[\frac{2k_BT}{\pi m}\right]^{1/2} \langle \delta(x-q) \rangle_A$$

$$= \left[\frac{2k_BT}{\pi m}\right]^{1/2} \frac{\int_{-\infty}^q \delta(x-q)e^{-\beta\frac{1}{2}m\omega^2 x^2} dx}{\int_{-\infty}^q e^{-\beta\frac{1}{2}m\omega^2 x^2} dx}$$
$$= \left[\frac{2k_BT}{\pi m}\right]^{1/2} \frac{\frac{1}{2}e^{-\beta E_a}}{\left[\frac{2k_BT}{\pi m}\right]^{1/2} \frac{\pi}{2}}$$

take this q to infinity so we can integrate analytically

 $V(x) = (1/2)m\omega^2 x^2$ 

$$k_{A\to}^{\rm HTST} = \nu_0 e^{-\beta E_a}$$

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Ea

x=q

#### **Vineyard Expression**

Vineyard, J. Phys. Chem. Solids, 3, 121 (1957)

Harmonic TST in 3N dimensions



 $\{v_i^{min}\} = 3N$  normal-mode frequencies at minimum  $\{v_i^{sad}\} = 3N-1$  nonimaginary frequencies at saddle

- Predicts rate constant for any T, using only the information at the minimum and the saddle point.
- $\bullet$  Typically a very good approximation below half of  $T_{\rm melt}$
- A.F. Voter, September, 2005 2AUK 55-8125, typical prefactor is 10<sup>12</sup> 10<sup>13</sup>

## **Effects of Anharmonicity**



#### **Infrequent Event System**



The system vibrates in 3N dimensional basin many times before finding an escape path. The trajectory finds an appropriate way out (i.e., proportional to the rate constant) without knowing about any of the escape paths except the one it first sees. Can we exploit this? A.F. Voter, September, 2005 LAUR-05-8125

### Accelerated dynamics concept

Let the trajectory, which is smarter than we are, find an appropriate way out of each state. The key is to coax it into doing so more quickly, using statistical mechanical concepts (primarily transition state theory).

With these accelerated dynamics methods, we can follow a system from state to state, reaching time scales that we can't achieve with molecular dynamics. Accelerated Molecular Dynamics Methods Hyperdynamics (1997)



Parallel Replica Dynamics (1998)



**Temperature Accelerated Dynamics (2000)** 

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