Home Work #2 - EE482

I- 
$$f_{MB} = exp(-(E-F_f))$$

For  $E-F_f>4kT$ 

Occupation

 $f_{FD} = \frac{1}{1+exp(E-F_f)}$ 

Probability

error = 
$$\frac{\text{App. }i}{\text{FMB}} - \frac{i}{\text{FD}} = \exp(-(E-Ef)) < \frac{3}{100}$$

$$=> E-EF> KTln(\frac{100}{3})=(3.51)KT$$

$$= > error = \left| \frac{\varphi_{P}}{\varphi_{B}} - \varphi_{FD} \right| = + \exp\left(+2(E - G\varphi)\right)$$

$$\frac{3}{100} = F - F f \left( -\frac{kT}{2} \ln \left( \frac{100}{3} \right) = -1.75 kT$$

Probability of being empty = 
$$1 - f_{FD}$$

$$= f_{NB} = \int 1 - \exp(-\frac{E - E_F}{kT}) \quad \text{for } E - E_F > 4 \text{NT} \text{ } \Omega$$

$$= \exp(\frac{E - E_F}{kT}) \quad \text{for } E - E_F < - 4 \text{NT} \text{ } \Omega$$

$$= \exp(\frac{E - E_F}{kT}) \quad \text{for } E - E_F < - 4 \text{NT} \text{ } \Omega$$

$$= \exp(-\frac{2(E - E_F)}{kT}) = \exp(-\frac{2(E - E_F)}{kT}) < \frac{3}{100}$$

$$= \sum_{E - E_F} = \exp(-\frac{E - E_F}{kT}) < \frac{3}{100}$$

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## 2(a) - Metal

If the crystal has Natoms per unit voling

the outmost band has 2N States, ie. 2 electrons

per atom (spin up and spin down). This bound

has only one electron per atom -> partially full

-> metal.

2(b) Here carrier (electron) Concentration n=N.

The out most bound of a metal is similar to the

conduction band.

$$n = \int_{E_0}^{\infty} g_c(E) f_{FO(6)} d6$$

therefore,

$$=\int_{E_0}^{E_f} \frac{4\pi}{h^3} (2me^*)^{3/2} (E_{-E_0})^{\frac{1}{2}} dE$$

$$N = \frac{4\pi}{h^3} \left(2me^{\frac{1}{2}}\right)^{\frac{3}{2}} \left(\frac{2}{3} \left(E_F - E_0\right)^{\frac{3}{2}}\right)$$

$$\frac{h^{3}}{2me^{*}}$$

$$\frac{h^{3}}{8\pi} \frac{2}{2me^{*}}$$

3 cal @ 300°k, for Germanium. ni= 2.4x10/3 cm3 9 NC= 1.0x1019 cm 3NV=5.4x10 cm Phospherus is a group V material and therefore a doner atom. Given that it's a shallow dopant, we can assume full ionizations i.e. N = 5x10 cm3 => N = N + P ~ Nd + P and because that 5xl0=Nd+>>n; P=1.152x10 cm-3  $EF = EC - KTIN(\frac{NC}{n}) = EC - KTIN(\frac{10^{19}}{50016})$ =EC-5.3KT = EC-0.14eV/

$$n = \frac{n_i^2}{P} = 1.152 \times 10^9 \text{ cm}^{-3}$$

$$EP = EV + KT \left( \frac{NV}{P} \right) =$$

$$EP = EV + KT | n(NV) = \frac{P}{P}$$

$$= EV + 0.026 eV | n(\frac{5.4 \times 10^{18}}{5 \times 10^{17}}) = EV + 0.062 eV$$

4 (a) 8 1- Charge Neutrality -> P+Nd+=n+NA 2- Equilibrium mass action -> np=ni2  $> ni^2 - n + (Nd^4 - Na^2) = 0$  $= > n^2 - (Nd^4 - NaT)n - ni^2 = 0$ (Nd-Na) + \(\((NJ^{+}-Na^{-})^{2}+4n^{2}\) (n-type  $E_{f} - E_{f}$ ;  $= KT \ln \left( \frac{n}{n} \right) = KT \ln \left( \frac{NJ - Na}{2n} + \sqrt{\frac{NJ - Na}{2n}} \right) + 1$ If we have full ionization, then, NJ-Na=NJ-Na by which we can rewrite the equations as functions of Nd-Na.

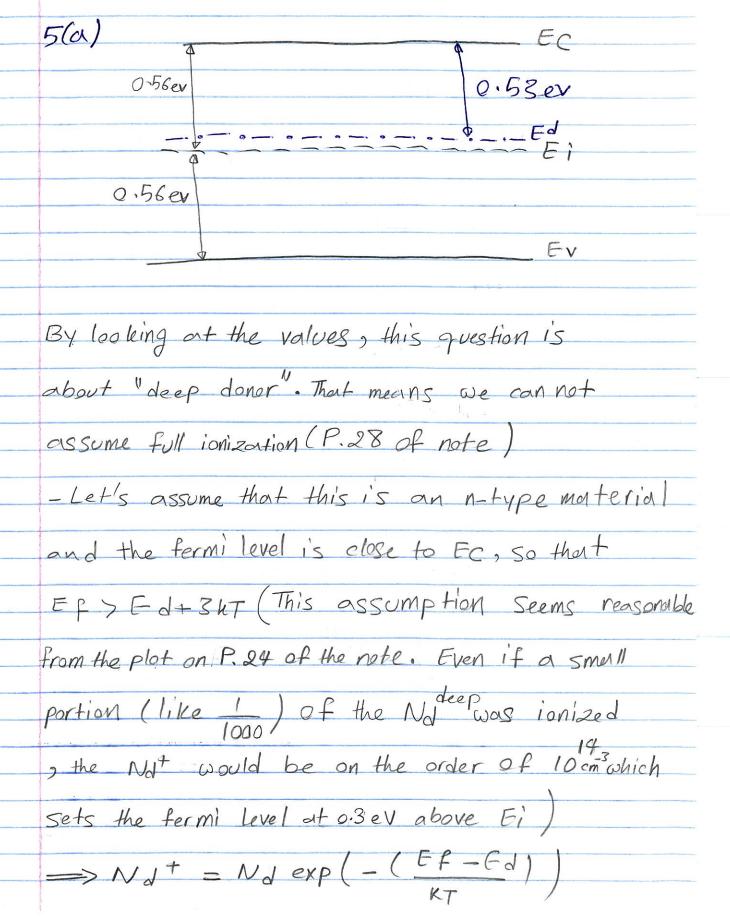
4 (b)
$$N_{d} = 10^{18} cm^{-3} (As), @750^{\circ}C = 1023^{\circ}k, n_{1} = 500^{18} cm^{-3}$$

$$= > n = \frac{Nd}{2} + \sqrt{Nd^{2} + 4n_{1}^{2}} = 5.52 \times 10^{18} cm^{-3}$$

$$P = \frac{n_{1}^{2}}{2} = 4.52 \times 10^{18} cm^{-3}$$

$$E f - E f_{1}^{\circ} = KT \ln \left( \frac{n}{n_{1}} \right) = KT \ln \left( \frac{5.52 \times 10}{5 \times 10^{18}} \right) = 0.1 \text{KMT}$$

$$= 0.1 \times 1023 \times 8.62 \times 10^{5} \text{ eV} = 8.72 \text{ meV}$$



- Charge Neutrality:

$$N = P + NJ^{+}$$

and  $n > p$  (for  $n - type$  material)

 $N = \exp\left(\frac{Ef - Gc}{kT}\right) = NJ \exp\left(\frac{EJ - Ef}{kT}\right)$ 
 $NC = \exp\left(\frac{EJ - Ef - Ef + EC}{kT}\right)$ 
 $NJ = \exp\left(\frac{EJ - 2Ef + EC}{kT}\right)$ 
 $NJ = \exp\left(\frac{EJ - 2Ef + EC}{kT}\right)$ 
 $NJ = \exp\left(\frac{NJ}{NJ}\right) = \frac{NJ}{NJ} = \frac{NJ}{NJ}$ 

Shallow Acceptor

$$5(e) \ Nd = 10^{17} \ cm^{-3} \ NA = 2 \times 10^{17} cm^{-3}$$
 $\Rightarrow P = NA - Nd^{+} = 10^{17} cm^{-3}$ 
 $= 10^{20} \ cm^{-3} = 10^{3} \ cm^{-3}$ 
 $= 10^{17} = 10^{3} \ cm^{-3}$