## Homework #4 Solutions, EE/MSE 486, Spring 2017

## **Problem 1**

From simulation result of problem 3 of homework 3, we can get the average oxidation rate for the first half hour,  $dx_0/dt=13.6$ nm/0.5hr=0.027um/h; Also for second hour the oxidation rate is  $dx_0/dt=(31.6$ nm-13.6nm)/(2hr-0.5hr)=0.012um/h.

From the Fig. 11 in the supplement notes for oxidation, we have for

$$\frac{dx_0}{dt} = \begin{cases} 0.027\mu m/h \\ 0.012\mu m/h \end{cases} \rightarrow \frac{D}{D^*} = \begin{cases} 4.3 \text{ for the first 30 mins} \\ 3.9 \text{ for the final 90 mins} \end{cases}$$

The above diffusion enhancement is for phosphorus, so we have

$$\frac{D}{D^*} = \frac{D_P}{D_p^*} = f_I^P \frac{c_I}{c_I^*} + f_V^P \frac{c_V}{c_V^*}, \quad where \ f_I^P + f_V^P = 1$$
(1)

Since the I/V recombine is assumed near equilibrium, so

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$$C_I * C_V = C_I^* * C_V^* \to \frac{c_V}{c_V^*} = \frac{c_I^*}{c_I}$$

$$\tag{2}$$

Combine (1) and (2), we can get

$$\frac{C_I}{C_I^*} = \begin{cases} 4.5 \text{ for the first 30 mins} \\ 4.1 \text{ for the final 90 mins} \end{cases}$$

For Arsenic,  $f_I^{As}$ =0.4, so we have

$$\frac{D_{As}}{D_{As}^*} = 0.4 \frac{C_I}{C_I^*} + 0.6 \frac{C_I^*}{C_I} = \begin{cases} 1.93 \text{ for the first 30 mins} \\ 1.79 \text{ for the final 90 mins} \end{cases}$$

For Antimony,  $f_I^{As} = 0.05$ , so we have

$$\frac{D_{AS}}{D_{AS}^*} = 0.05 \frac{C_I}{C_I^*} + 0.95 \frac{C_I^*}{C_I} = \begin{cases} 0.44 \text{ for the first 30 mins} \\ 0.44 \text{ for the final 90 mins} \end{cases}$$

In Sentaurus simulation, the equilibrium interstitial concentration at 1000C is  $7.7*10^{11}$  cm<sup>-3</sup>.

The output of timedependent interstitial concentration during oxidation is shown below:



For first 30 minutes oxidation at 1000 C, the average interstitial concentration is about  $2.2*10^{11}$  cm<sup>-3</sup>, and  $C_I/C_I^* = 2.85$ .

For final 90 minutes oxidation at 1000 C, the average interstitial concentration is about  $1.6*10^{11}$  cm<sup>-3</sup>, and  $C_I/C_I^* = 2.08$ .

## **Problem 2**

According to the Range Statistics plot, we can get Rp and  $\Delta$ Rp for SiO2 and photoresist for B implantation with 5keV.

For SiO2, Rp=18 nm, and  $\Delta$ Rp=11nm

For photoresist, Rp=40nm, and  $\Delta$ Rp=11nm

The relation between penetrated dose and film thickness is

$$Q_P = \frac{Q}{2} erfc(\frac{x_m - R_P}{\sqrt{2}\Delta R_P})$$

In this problem,  $Q=10^{14}$  cm<sup>-2</sup>, and  $Q_P=2*10^{12}$  cm<sup>-2</sup>.

$$erfc\left(\frac{x_m - R_P}{\sqrt{2}\Delta R_P}\right) = 0.04 \quad => \left(\frac{x_m - R_P}{\sqrt{2}\Delta R_P}\right) = 1.45$$

For SiO2, the above equation can be solved to get  $x_m$ =40.5nm

For photoresist, the above equation can be solved to get  $x_m=62.5$ nm

The Sentaurus input is as follows:

line x location=0 spacing= 0.001 tag=SiDevTop line x location=1 spacing=0.01 line x location=5 spacing=0.01 tag=SiDevBot region silicon xlo=SiDevTop xhi=SiDevBot init concentration=1e14<cm-3> field=Phosphorus wafer.orient=100

deposit material=photoresist type=isotropic thickness=0.0974 grid remesh

implant Boron energy=5<keV> dose=1e14<cm-2> tilt=7

strip photoresist diffuse temperature=750<C> time=1<s> #diffuse temperature=1000<C> time=0.5<s>

struct tdr=p1 select z=BTotal layers

The simulated thickness to obtain required penetrated dose is 42.5 nm for oxide and 97.4 nm for photoresist.

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Edges: 665 Faces: 0 Volumes: 0				
Writing file	p1_fps.t	tdrdone		
select Z= "E	Total"			select
layers				layers
{ To { 0.00000000	p 00000e+00	Bottom 5.000000000000e+00	Integral 2.035682579458e+12	Material } Silicon }
Summary: Anneal time reaction/m diffusion	stepping: mechanics steps=14	: steps=14 minimum=1. minimum=1.0000e-04s	.0000e+38s maximum=1 s maximum=0.470762s	L.0000e+38s
Elapsed Time User Time: CPU Time:	e: 00:00 00:00 00:00	0:25 0:24 0:22		

## **Problem 3**

If the beam is aligned with the channels, then nuclear stopping will be minimized and electronic stopping will dominate. The range will be

$$R = \frac{1}{N} \int_0^E \frac{dE}{S_e(E)} = \frac{1}{N} \int_0^E \frac{dE}{k\sqrt{E}}$$

The value of k is given by  $k = 0.2 \times 10^{15} eV^{1/2} cm^2$  while  $N = 5 \times 10^{22} cm^{-3}$  is the atom density in silicon, so that the range is given by

 $R \approx 20 \sqrt{E} Angstroms \approx 282 nm$ 

This is about four times the range from the range tables (Rp=74 nm and delta Rp=31nm), because there is a nuclear stopping component which contributes along with the electronic stopping. However, the calculation does indicate the range of a well-channeled ion. For a  $10^{14}$ cm<sup>-2</sup> dose implantation, the implantation profile with 50% ions channeled is shown below:

